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Preliminary communication

PHOTOREACTION OF O-QUINODIMETHANETRICARBONYLIRON WITH PENTACARBONYLIRON

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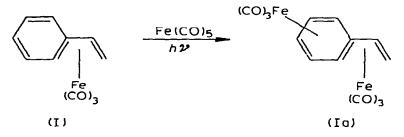
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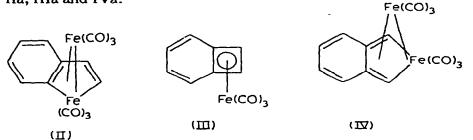
Summary

Irradiation of (*tetrahapto-o*-quinodimethane)tricarbonyliron in the presence of $Fe(CO)_5$ gives three isomeric $C_8 H_8 Fe_2 (CO)_6$ complexes.

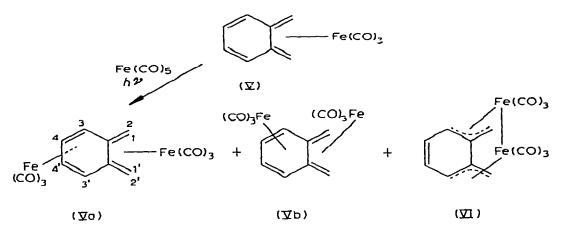
We recently described the first example of a photochemical reaction in which a *tetrahapto*-tricarbonyliron complex, I, of the 8π -electron conjugated type gives the bis(*tetrahapto*) derivative, Ia on reaction with pentacarbonyliron [1]. Later we showed that similar reactions can be carried out with the



related 8π -electron systems of ferraindentricarbonyliron, II [2], benzocyclobutadienetricarbonyliron, III [3], and ferra-2H-indenetricarbonyliron, IV [3], leading to the formation of the respective bis(*tetrahapto*) complexes IIa, IIIa and IVa.



These results, and in particular the formation of IVa from IV, suggested that o-quinodimethanecarbonyliron, V [4], should by analogy form the corresponding bis(tetrahapto) derivative under similar conditions.



The photochemical reaction [3] of V with $Fe(CO)_5$ has been found to give three isomeric $C_8 H_8 Fe_2$ (CO), complexes in comparable yields: Va, bis-(tetrahapto)-o-quinodimethane-trans-bis(tricarbonyliron), red crystals, m.p. 94–96° (decomp.); Vb, bis(tetrahapto)-o-quinodimethane-cis-bis(tricarbonyliron), purple crystals, m.p. 95° (decomp.) and VI, bis(trihapto)-oquinodimethane-hexacarbonyldiiron, red crystals, m.p. 126-127° (decomp.). Formation of Va and Vb from V involves the expected diene-type coordination of the residual four π -electron system, while that of VI involves a redistribution of the π -electron density in the 8π -electron-conjugated system.

The structures of V_{ϵ} , Vb and VI follow from their physical, analytical and spectroscopic properties*.

Va: m/e 384 (M⁺). IR: 1962, 1977, 1983, 2000, 2041, 2055 cm⁻¹ (v(Fe-CO)) NMR (τ) : 3.26m (H-4), 6.61m (H-3), 8.29d (H-2), 9.80d (H-1), $J_{12} \sim 3.0, J_{34} \sim 6.4, J_{34'} \sim 1.6, J_{44'} \sim 4.5$ Hz.

Vb: m/e 356 [($\dot{M} - CO$)⁺]. IR: 1930, 1982, 1990, 2044 cm⁻¹ (ν (Fe-CO))^{*} NMR (τ): 3.95m (H-4), 7.02m (H-3), 9.37d (H-2), 9.70d (H-1), $J_{12} \sim 5.1$, $J_{34} + J_{34'} \sim 7.2$ Hz. The assignment of the cis-structure to Vb is supported! by the following: (a) Vb is more polar and less stable than Va, (b) Vb is converted readily on standing in solution to complex VI, which by virtue of its iron-iron bond involves a cis Fe-Fe relationship.

VI: m/e 384 (M⁺). IR (Nujol): 1958, 1964, 1980, 1998, 2020, 2061 cm⁻¹ $(\nu(\text{Fe-CO}))^{***}$ and 752 cm^{-1} $(\nu(\text{C=C}))$. NMR (τ) : 3.63m (H-4), 5.49m (H-3),

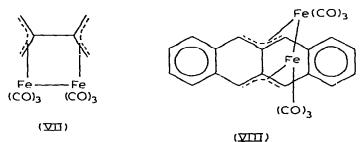
[&]quot;On column chromatography (Flonsil), p.e. 30-40, the products were eluted in the order: (a) unreacted V, (b) Va, (c) VI and (d) Vb. All the new compounds gave satisfactory elemental analysis. Unless otherwise indicated, IR spectra refer to bexane solutions, and NMR spectra to CDCl, solutions,

The Fe-CO absorptions are analogous to those of styrenebis(tricarbonyluron), for which the trans relationship has been established [5]. *** The pattern of the $\nu(CO)$ absorptions is characteristic for a bis π -allene structure, cf. refs. 6

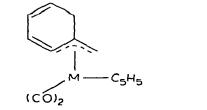
and 7.

8.45d and 8.50d (H-2 and H-1), $J_{12} \sim 3.1$, $J_{34} \sim 6.6$, $J_{34'} \sim 1.0$, $J_{44'} \sim 10.3$ Hz^{*}

Complex VI belongs to the family of bis- π -allylene iron complexes for which the parent complex VII [6] and the naphthacene derivative VIII [7] have been reported. It is also structurally related to the π -benzyl- π -cyclopentadienylcarbonyl complexes IX, of molybdenum [8] and tungsten [8c]. Its identification as the remaining C₆ H₃ Fe₂ (CO)₆ isomer was based inter alia on (a) the presence of an uncoordinated double bond in the ligand (vide IR), (b) the presence of an Fe-Fe bond, as indicated by the mass spectrum (strong peak at m/e 112), (c) the sp^2 character of the exocyclic carbon bonds, as shown by ¹³C-¹H NMR J(¹³C-H) ~ 157 Hz), which excludes any structure involving C-Fe σ bonds, and (d) the formation of α, α -dibromoxylene on bromine degradation, consistent with the formation of 2,3-bis(bromomethyl)-1,4-dibromobutene on similar bromination of VII [6a].







(IXa) $M = MO, R = H \text{ or } CH_3$ (IXb) $M = W, R = CH_3$

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References

- 1 R. Victor, R. Ben-Shoshan and S. Sarel, J. Org. Chem., 37 (1972) 1930.
- 2 (a) R. Victor, R. Ben-Shoshan and S. Sarel, Chem. Commun., (1971) 1241.
- (b) Unpublished results from this laboratory.
- 3 R. Victor and R. Ben-Shoshan, Chem. Commun., (1974) 93.
- 4 W.R. Roth and J.D. Meier, Tetrabedron Lett., (1967) 2053.
- 5 F.H. Herbstein and M.G. Reisner, Chem. Commun., (1972) 1077.

^{*}This value for the exocyclic geminal bydrogen coupling is in accord with that for a related trihapto complex, the π -benzyl complex [Xa, $R \approx H (J \sim 3.0 \text{ Hz})$ [8a].

- 6 (a) A. Nakamura and N. Hagihara, J. Organometal. Chem., 3 (1965) 480.
 (b) R. Ben-Shoshan and P. Pettit, Chem. Commun., (1968) 247.
 7 R.A. Bauer, E.O. Fischer and C.G. Kreiter, J. Organometal. Chem., 24 (1970) 737.
- 8 (a) R.B. King and A. Fronzaglia, J. Amer. Chem. Soc., 88 (1966) 709.
 (b) F.A. Cotton and M.D. LaPrade, ibid., 90 (1968) 5418.

- (c) F.A. Cotton and T.J. Marks, ibid., 91 (1969) 1339. 9 Ci.: K. Ebrlich and G.F. Emerson, J. Amer. Chem. Soc., 94 (1972) 2464 and. ref. therein.